

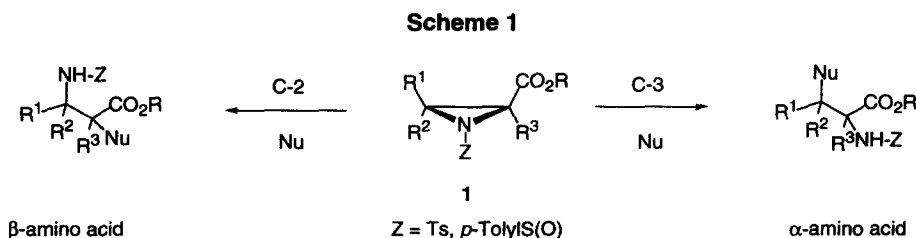
AZIRIDINE 2-CARBOXYLATE ESTER MEDIATED ASYMMETRIC SYNTHESIS OF α -ALKYL β -AMINO ACIDS

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Summary: The highly stereoselective ring opening of N-tosylaziridine 2-carboxylate esters with LiAlH_4 followed by oxidation of the ensuing *syn* alcohols results in a highly efficient 4 step asymmetric synthesis of α -methyl β -amino acids from N-sulfinylaziridine 2-carboxylate esters. © 1997 Elsevier Science Ltd.

The asymmetric synthesis of α -alkyl β -amino acids is of current interest because they are constituents of biologically active natural products¹ and precursors of the β -lactam class of antibiotics. Furthermore, substitution of α -amino acids by β -amino acids is useful in the preparation of peptide analogs with increased activity and enzymatic stability.² Methods for their synthesis³ include the Michael addition of chiral secondary amines to α,β -unsaturated esters,⁴ the homologation of α -amino acids,⁵ alkylation of enolates derived from perhydropyrimidin-4-ones⁶, additions to imines,⁷ and enzymatic resolution.⁸ Many of these procedures are multi-step, affording the product in low overall conversions or requiring separation of diastereoisomers. For these reasons the development of more efficient methods for the asymmetric synthesis of this important class of β -amino acids is highly desirable.

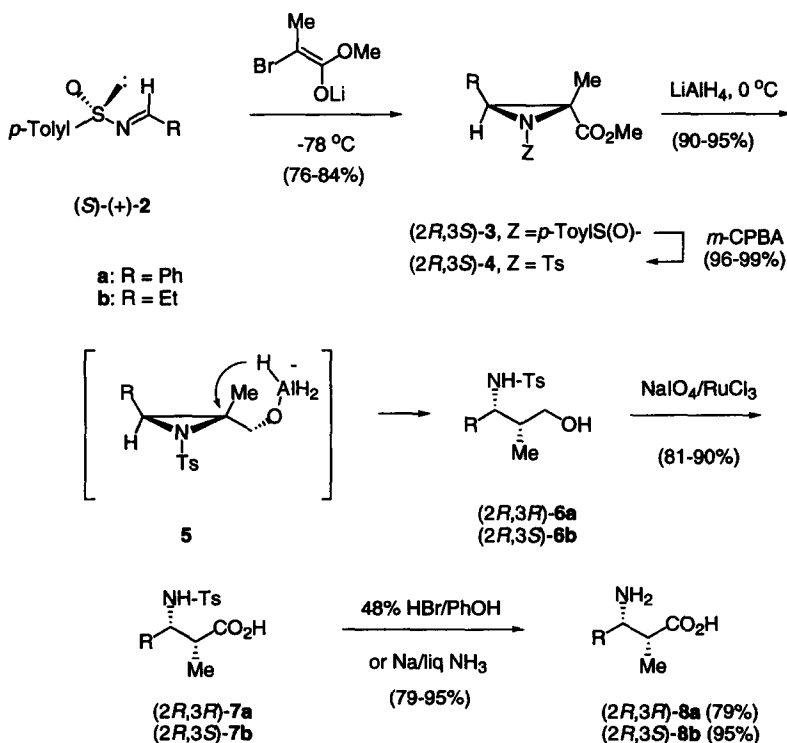
N-Activated aziridine 2-carboxylate esters **1** are useful building blocks for the asymmetric construction of α -amino acids because they undergo stereocontrolled ring-opening at C-3 with inversion of configuration to give β -substituted α -amino acid (Scheme 1).^{9,10} Indeed, recent studies in our laboratory have demonstrated the application of *cis*-N-sulfinylaziridine 2-carboxylate esters **1** ($\text{Z} = \text{ArS(O)-}$) in highly stereoselective asymmetric synthesis of α -amino acids,¹⁰ α -methyl¹¹ and β -substituted α -amino acids,¹² the antibiotic thiamphenacol,¹³ β -hydroxy α -amino acids,¹⁴ β -hydroxy α -methyl α -amino acids,¹¹ the protein kinase C inhibitors *D-erythro* and *L-threo*-sphingosine,¹⁴ and



the cytotoxic antibiotic (*R*)-(-)-dysidazirine.¹⁵ However, to prepare β -amino acids from **1**, requires ring-opening at C-2 which is relatively rare.¹⁶ On the other hand, Tanner¹⁷ and others¹⁸ have shown that C-2 aziridine ring-opening occurs in the corresponding aziridino alcohols **1** ($\text{CO}_2\text{R} = \text{CH}_2\text{OH}$). We make use of this strategy in the first asymmetric synthesis of α -methyl- β -amino acids from aziridine-2-carboxylate esters.

2-Methyl-N-sulfinylaziridines **3a-b** were prepared as previously described by addition of the lithium enolate of methyl α -bromopropionate to sulfinimines (*S*)-**2a-b** (Scheme 2).¹⁰ (*S*)-(+)-*N*-(propylidene)-*p*-toluenesulfinamide (**2b**) was prepared in 67% yield by reaction of (1*R*,2*R*,5*R*)-(-)-menthyl (*S*)-*p*-toluenesulfinate (Andersen reagent) with 1.3 equivalents of LiHMDS and 1.1 equiv. *n*-propanal at -78 °C as present earlier.^{19,20} Following separation of the diastereomeric *E/Z*-aziridines by flash column chromatography (+)-(2*R*,3*S*)-**3a** and (+)-(2*R*,3*S*)-**3b** were isolated in 84 and 76% yield, respectively.²⁰ Oxidation of the *N*-sulfinyl aziridines **3** was accomplished using 2.0 equiv. of 60% *m*-chloroperbenzoic acid (*m*-CPBA) in CHCl_3 to give the *N*-tosyl derivatives **4** in nearly quantitative yield. Reduction of **4** at 0 °C with 1.0-1.2 equiv. of lithium aluminum hydride (LiAlH_4) afforded exclusively the *syn* alcohols (2*R*,3*R*)-**6a** and (2*R*,3*S*)-**6b** in 90-95% yield.²⁰ 2-Methyl-3-(*N*-tolylamino)-3-phenyl-1-propanol (**6a**) has been prepared in racemic form as a 1:1 *syn:anti* mixture

Scheme 2



via the hydroboration of 2-methyl-1-(N-tolylamino)-1-phenyl-2-propene,²¹ and (2*S*,3*R*)-2-methyl-3-(tosylamino)pentanol (**6b**) was prepared in 10 steps from L-aspartic acid.²² The fact that the reductive ring-opening of **4** occurs at C-2 with inversion of configuration strongly supports the suggestion by Tanner et al.¹⁷ that hydride is delivered intramolecularly via the aluminum hydride complex of the intermediate aziridino alcohol; e.g. **5**.

Next the amino alcohols **6** were oxidized to the β -amino acid derivatives **7** in good to excellent yield by treatment with 4.4 equiv. of NaIO₄ and a catalytic amount of RuCl₃. Attempts to remove the tosyl group in **7a**, as previously described,¹² by heating with 48% HBr/PhOH gave uncharacterizable materials. However, reduction with Na/liq. NH₃ afforded (2*R*,3*R*)-(+)-2-methyl-3-amino-3-phenylpropanoic acid (**8a**)²² in 79% isolated yield. The tosyl group in **7b** was removed uneventfully with 48% HBr/PhOH to give (2*R*,3*S*)-(-)-2-methyl-3-aminopentanoic acid (**8b**)²³ in 95% isolated yield. The fact that the **8a** and **8b** were obtained as single diastereoisomers (¹H NMR of the crude mixtures) indicates that removal of the tosyl group occurred without epimerization at the C-3 nitrogen stereogenic center. The crude amino acids were purified using a Dowex 50X8-100 ion exchange resin and isolated as the hydrochloride salts by dissolving in conc. HCl and concentrating. Their properties were identical to authentic samples and obtained in >97% ee.²²⁻²⁴ Amino acid **8b** is the enantiomer of the β -amino acid found in the marine cytotoxic antitumor desipeptides dolastatin D,^{1e} **10** and **11**.^{1a} This enantiomer can be readily prepared starting from (*R*)-(-)-**2**.¹⁹

In summary, a short, highly efficient four-step stereoselective asymmetric synthesis of α -alkyl β -amino acids **8a** and **8b** in 57-81% overall yield from N-sulfinylaziridine-2-carboxylate esters **3a** and **3b** is described.

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20. Selected physical properties: (*S*)-(+)-**2b**: oil, $[\alpha]^{20}_{\text{D}} +385$ (c 2, CHCl₃); *E*-(*Ss*,*2R*,*3S*)-(+)-**3b**: oil; $[\alpha]^{20}_{\text{D}} +72.0$ (c 0.30, CHCl₃); *E*-(*2R*,*3S*)-(+)-**4b**: mp 84-86 °C, $[\alpha]^{20}_{\text{D}} +24.4$ (c 0.55, CHCl₃); *syn*-(*2R*,*3R*)-(+)-**6a**: mp 159-61 °C, $[\alpha]^{20}_{\text{D}} = +26.1$ (c 1.0, CH₃OH); *syn*-(*2R*,*S*)-(-)-**6b**: mp 73-75 °C, $[\alpha]^{20}_{\text{D}} -35.0$ (c 0.60, CHCl₃); *syn*-(*2R*,*3R*)-(+)-**7a**: mp 144-46 °C, $[\alpha]^{20}_{\text{D}} = +49.1$ (c 1.1, CHCl₃); *syn*-(*2R*,*3S*)-(-)-**7b**: oil; $[\alpha]^{20}_{\text{D}} -36.5$ (c 0.62, CHCl₃).
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